

Can tolerate maybe 5 px blur?

10 Mpx ~ 3750 X 2750

 $0.1 \text{ m} / 3750 = 2.6 \text{ e-5} = 0.000026 \text{ m/px} = 26 \mu\text{m/px}$ 

5 px = 1.3 e-4 m = 0.00013 = 0.13 mm estimated acceptable object displacement x time t = x/velocity

1.3e-4 m / (0.5 m/s) = 2.6e-4 seconds

 $2.6e-4 \text{ sec} = .00026 \text{ sec} = 260 \,\mu\text{sec} = 1/3750 \,\text{Very short. Can your camera do this?}$  5/3750 = 0.0013 = 0.13% of image width

Do this analysis for each image. Motion blur is surprisingly common and annoying.

If unacceptable, increase time resolution= shorter exposure time

Increase shutter speed

Max is 1/10,000? 0.1 msec, 100 μsec? At best.

High speed camera 30,000 fps  $\sim$  3 x 10-5 sec = 30  $\mu$ sec

Freeze the flow with short light source (won't work for light emitting fluids, i.e. flames)

Strobe, camera flash  $\sim$  10-5 or -6 sec = 1-10 µsec

Pulsed laser 3x10-9 sec = 3 nsec or less

Good resource for high speed photography: http://www.hiviz.com/index.html

## **SPECIFIC FV techniques**

Boundary techniques. Boundary between 'seeded' and unseeded fluid.

Choice depends on physics desired

I DYES Today. Mostly in water.

2 Aerosols Particles. Mostly in air for boundary effect.

In this class, often visualization technique determines physics examined, but usually physics are determined by system under study, and FV technique applied should not disturb the flow/physics

## I Dye Considerations:

- 1)Want dye to NOT disturb flow
- 2)Want dye to show up HIGH VISIBILITY
- 3) Special techniques
- 1) Not Disturb flow "How?"

  Minute paper -Groups

## **Answers:**

- Match flow speed when injecting
- Use small ports, minimize volume injected,
- Consider location of injection; reveals different physics http://media.efluids.com/galleries/laminar?medium=113



by Henri Werlé, at
ONERA = NASA of France
Master of colored dye
streams

Avoid injection altogether: Coat object with alcohol-dye mixture, let dry, then tow in tank. Shows vorticity layer, wake, boundary layer

Or coat short strings on a rake. OK for low speed, short run times

- Match fluid properties between dye and medium
- Density

Temperature

Viscosity

Surface tension (match intermolecular forces)

- Minimize chemical reactions (unless needed)
- Diffusion coefficient

N.J. Mueschke et al., "Measurements of molecular mixing in a high-Schmidt-number Rayleigh-Taylor mixing layer," *Journal of Fluid Mechanics* 632, J. Fluid Mech. (UK) (2009): 17-48.

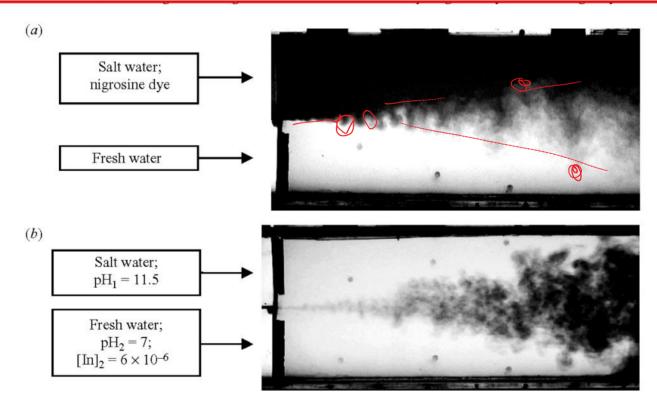


FIGURE 4. Photographs (contrast enhanced for visualization) of the buoyancy-generated mixing layer in a typical water channel experiment. (a) Nigrosine dye was added to the top stream. (b) Phenolphthalein was added to the bottom stream, which changes to its pink form as the two streams molecularly mix (here, "pink" is shown as dark regions within the mixing layer).

Ph indicator, shows where mixing got to molecular level.

Tough to match all these properties- Dye properties are different from ambient fluid. To match density, try a premix. For food dye in water, premix dye (dense, sinks in water) and isopropyl alcohol (floats) to get neutral buoyancy in water The concentration gradient between dyed and undyed fluid may cause dye to diffuse too rapidly, misleading when studying mixing. Turbulence also causes fast diffusion, making visualization of the overall flow structure difficult. Try some milk or latex paint to slow diffusion. Famous example: Cloud tank was invented by Douglas Trumball to make realistic clouds in 'Close encounters of the third kind' (1980's sci fi). Used many times since: http://www.youtube.com/watch?v=2Ps0iXwS60E More info in Special Effects article http://www.americanheritage.com/articles/magazine/it/2007/1/2007 1 10.shtml